

**Synthesis, Spectral Characterisation And Antimicrobial Studies Of Pyrazole Derivatives****Neha Sharma****Research Scholar****Department of Applied Chemistry****Maharaja Ranjit Singh Punjab Technical University****Bathinda****(Received:20May2020/Revised:13June2020/Accepted:22June2020/Published:28June2020)****Abstract**

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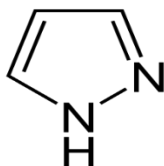
Nivel 1-(4-substitutedphenyl)-3-(4-substituted phenyl)-prop-2-en-1-ones were synthesized from a Claisen-Schmidt reaction of 4-substituted benzaldehyde with several acetophenone derivatives. Subsequently, the cyclocondensation by reaction of chalcones with hydrazine 5-(4-substitutedphenyl)-3-(4-substituted phenyl)-1H-pyrazole derivatives. Several of these compounds were screened for the ability to inhibit *Staphylococcus aureus* (S.aureus) as gram positive bacteria and *Escherichia coli* (E. Coli) as gram negative bacteria. *Candida albicans* & *Aspergillus niger* were strains used for antifungal activity.

**Keywords:** Chalcones, Pyrazole, Antibacterial, Antifungal, IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR, MASS.

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**Introduction**

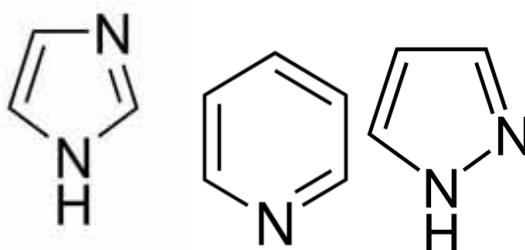
Pyrazoles are five member ring heterocyclic compounds, have some structural features with two nitrogen atoms in adjacent position. Among the two nitrogen atoms; one is basic and the other is neutral in nature. These are aromatic molecules due to their planar conjugated ring structures with six delocalized  $\pi$ -electrons. The aromatic nature arises from the four  $\pi$  electrons and the unshared pair of electrons on the -NH nitrogen<sup>[1]</sup>.

**Comparison of Basicity Among Pyrazole, Imidazole & Pyridine**

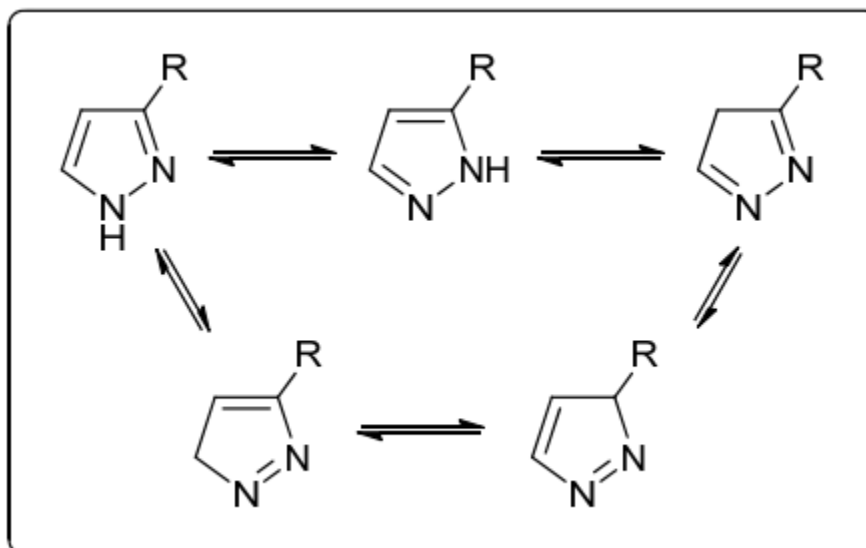
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Basicity of compound is based on electron pair availability on atoms. Among of these heterocyclic compounds containing a nitrogen atom as a hetero atom in a ring; there is basicity

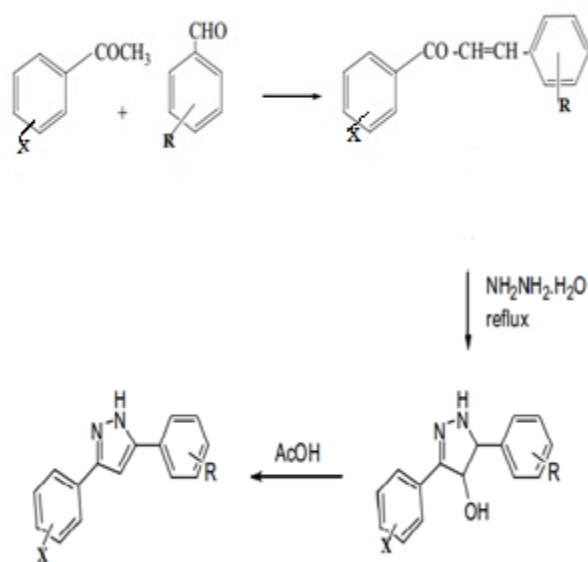
depends upon the ability to donate the lone pair of nitrogen atom. Both imidazole & pyrazole containing two nitrogen atom in their ring. But pyrazole form a dimer so, its lone pairs are busy in intramolecular hydrogen bonding. This type of dimer structure is not possible in imidazole. So Its lone pairs are available to donate electrons so imadazole more basic than pyrazoles. But in comparison with pyridine the more basic heterocyclic compound is imidazole because the presence of additional N, which is more electronegative the C, makes the whole ring more electron rich, which means more basic . So, we can say that imidazole is most basic among of these three heterocyclic compounds .but in comparison with pyridine & pyrazole, there is pyrazole is less basic than pyridine because in pyrazole lone pair of electrons on N is involved in resonance whereas in pyridine this is not possible. So Pyrazole is least basic or weak base.



For pyrazoles in which two carbon neighbouring the nitrogen atom in ring have different substituent. Five tautomeric structures are possible are given as following:



Pyrazole is synthesised from chalcones which were obtained by condensation of acetophenone with aromatic aldehydes derivatives in presence of suitable condensing agent. Chalcones have been used as intermediate for the preparations of compounds having therapeutic value. They undergo a various chemical reactions and are found useful in synthesis of variety of pyrazole derivatives [2,3]. Pyrazole derivatives are the subject of many research studies due to their widespread potential biological activities such as antimicrobial [4], antiviral [5], antitumor[6,7], antidepressant[8], insecticides[9] and fungicides [10].In the view of the varied biological and pharmacological applications, we have planned to synthesize some pyrazole derivatives from various chalcones and test their antibacterial activity and antifungal properties.



**Scheme 1**

## Experimental

### Material And Instruments

All chemicals and reagents were obtained analytical grade and used without further purification. The IR spectra (in KBr pellets) were recorded on a Perkin -Elimer 400 FTIR spectrometer (Germany). <sup>13</sup>C NMR spectra were recorded in Bruker Avance II 400 NMR (Fallanden), NMR spectrometer using TMS as an internal standard. The MASS Spectroscopy were recorded in WATERS, Q-TOF micromass LC-MS (UK). The compounds were analyzed for elemental analysis and the percentages of elements were found to be very near that of the

calculated values by using software Accelrys Draw 4.1 . Physical data of the compounds are recorded in Table-1

### Synthesis of chalcone(I)

Benzaldehyde derivative (0.01 mol) and acetophenone (0.01 mol) were dissolved in ethanol (25 mL) Sodium hydroxide solution 10% (25 mL) was added slowly and the mixture stirred for 9 hrs then it was poured into 400 mL of water with adding few drops of HCl and left overnight in Refrigerator. The precipitate obtained was filtered, washed and recrystallized from ethanol.

### Synthesis of Pyrazole derivatives (II)

A mixture of chalcone (0.02 mol) and hydrazine hydrates (0.02 mol) in glacial acetic acid or ethanol (25 mL) was refluxed for 9-11 hrs, and then the reaction mixture was poured into ice water (50 mL). The precipitate obtained was filtered, washed and recrystallized from methanol or ethanol.

Cmpd No.	Molecular Formula	Mol. wt.	Solubility	Elemental Analysis (Calculated)					
				%C	%H	%O	%N	%Cl	%Br
I	C <sub>15</sub> H <sub>10</sub> BrN <sub>3</sub> O <sub>2</sub>	342.995	CDCl <sub>3</sub>	52.25	2.93	9.30	12.21	-	23.22
II	C <sub>16</sub> H <sub>13</sub> ClN <sub>2</sub> O	284.74	CDCl <sub>3</sub>	67.49	4.60	5.62	9.84	12.45	-
III	C <sub>15</sub> H <sub>11</sub> ClN <sub>2</sub>	254.061	CDCl <sub>3</sub>	70.73	4.35	-	11.00	13.92	-

Table -1 Physical Parameters and Elemental Analysis of Synthesized Compounds

### Spectral analysis of synthesised Compounds

(a) **5-(4-bromophenylphenyl)-3-(4-nitrophenyl)-1H-pyrazole:** In the <sup>13</sup>C NMR spectrum of compound pyrazole derivative a confirmed pattern was observed peak signals. The characteristics signal corresponding to sp<sup>3</sup> carbon(C-NH) in pyrazole ring was observed at δ 47.48 ppm. The characteristics signal obtained by vinyl carbon(C=C & C=N) in pyrazole ring was characteristics at 124.7 and 130.17 ppm. The characteristics downfield shift due to aromatic ring attached to the pyrazole ring was observed in range of 120- 175 ppm.

(b) **5-(4-chlorophenyl)-3-(4-methoxyphenyl)-1H-pyrazole:** The characteristics band at 3023 cm<sup>-1</sup> attributing to Ar-H stretching vibrations. Band due to C=C stretching vibrations was observed in the range of 1513-1614 cm<sup>-1</sup>. The band observed at 2939 cm<sup>-1</sup> was characteristic of C-H stretching vibrations. The band due to stretching vibration of C=N group was observed at 1658 cm<sup>-1</sup>. A band at 3300 cm<sup>-1</sup> was observed due to N-H stretching vibrations. The band due to stretching vibration of N=CH group was observed at 1614 cm<sup>-1</sup>. The band due to stretching vibration of O-CH<sub>3</sub> group

was observed at  $1423\text{ cm}^{-1}$ . The band due to stretching vibration of C-Cl group was observed at  $825\text{ cm}^{-1}$ .

**(c) 3-(4-chlorophenyl)-5-phenyl-1H-pyrazole:** In the  $^{13}\text{C}$  NMR spectrum of compound pyrazole derivative a confirmed pattern was observed peak signals. The characteristics signal corresponding to  $\text{sp}^3$  carbon (C-NH) in pyrazole ring was observed at  $\delta$  52.64 ppm. The characteristics signal obtained by vinyl carbon (C=C & C=N) in pyrazole ring was characteristics at 126.64 ppm and 142.42 ppm.

### **Biological Assay Of The Synthesized Products**

Method of Antimicrobial Analysis

Methodology-The evaluation of the anti microbial effect of Synthesized Pyrazole was done by using Agar Well Diffusion method in the Department of microbiology, ISFAL (A unit of ISF College), Moga Punjab (INDIA).

### **Preparation of media and media plates**

Antibiotic Assay Medium No. 11 (30.5 gm/1000ml of distilled water) was dissolved and added in a conical flask. Then the flask was plugged with cotton and autoclaved for complete sterilization. The sterilized media was poured in sterile petri dishes aseptically in a laminar flow. After solidifying of Agar plates (nearly about 15 to 20 minutes) they were kept inverted in incubator at  $35\pm 2^\circ\text{C}$  for overnight for checking any contamination. The ready Agar plates then transferred in zip seal plastic cover and kept in a cold room.

### **Procurement of cultures**

The pathogenic strains of different species of *E.coli* (MTCC-1687) and *Staphylococcus aureus* (MTCC-737) bacteria and *Aspergillus niger* (MTCC-282), *Candida albicans* MTCC- 227 Fungus were procured from Department of Microbiology I.S.F. college of Pharmacy, Moga The cultures were in freeze dried form (i.e. in dormant state). So, their revival was necessary. For this 100 ml nutrient broth medium was made and transferred in five small conical flasks (of quantity 100ml) 20ml each. The flasks were capped with cotton plug and autoclaved at  $121^\circ\text{C}$  for 15 minutes at 15 lb pressure per square inch.

### **Spreading**

For isolation of micro- organisms in pure form without contamination, streaking was done on solid media plates by applying a microbial culture with a loop to the surface of Agar in a petri plate and spreading them with a sterile spreader. Already prepared solid media plates were used

for streaking process. A drop of previously made broth cultures of *E.coli*, *Staphylococcus aureus* *Candida albicans* and *Aspergillus niger* respectively was added at one edge of the two sets each of four agar plates and the spreading of cultures was done with sterilised spreader. Each time the spreader was sterilised on the burner flame and cooled in to the edge of agar in the respective plate. The spread of 16 culture plates, each set of 4 loaded each of *E.coli*, *Staphylococcus aureus* *Candida albicans* and *Aspergillus niger* respectively .

#### **Loading Of The Plates And Measurement Of Zone Of Inhibition**

By using sterile cavity cork borer of 8mm size, wells were made in the centre of each of incubated culture plate to enable the introduction of the test sample and standard control. With the help of micropipette 100µl of concerned sample of aqueous were introduced into well of each plate streaked with different bacterial and fungal stains of *E.coli*, *Staphylococcus aureus* *Candida albicans* and *Aspergillus niger* respectively. For comparison one plate each for *E.coli* and *Staphylococcus aures* was loaded with Ampicillin trihydrate and for *Aspergillus niger*, *Candida albicans* with Ketoconazole. Then the plates were allowed to stand by for 30 minutes and were incubated for a time period of 24 hrs at the temperature of 37°C. The zone of inhibition was examined and measured with the help of antibiotic zone read. Antibacterial activity of the Pyrazole derivatives have been carried out against several types of bacteria such as, *E. coli*; and *S. aureus*; by taking standard drug Ampicillin trihydrate ;& Antifungal activity have been carried out against *C.albicans* and *A.niger* by using Ketoconazole as a standard drug using nutrient agar medium by well diffusion method[11-14]. All compounds were suspended in aqueous solutions in different concentrations ranged from 10-100 mg/mL, the results are expressed on MIC (minimal inhibitory concentration), solvent blanks were run against each test organism in all assays and the experimental biological data is given in table:

Comp. 100 mcg/ml	Zone of Inhibition (mm)			
	Gram positive	Gram negative	Anti Fungal	
	S.aureus	E.coli	C.albicans	A.niger
<b>P1</b>	12.34	-	14.52	11.61
<b>P2</b>	20.17	-	17.16	13.24
<b>P3</b>	18.56	9.24	18.11	19.14
<b>Standard drug</b>	23.76	19.24	25.48	21.34
<b>Solvent control</b>	-	-	-	-

**Table 2: Antimicrobial activity of synthesized compounds**

#### **Results And Discussion**

All synthesized compounds as well as the reactions that carried out were characterized and

monitored by elemental analysis, FTIR and  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, MASS and they all gave satisfactory results as shown in Tables 1 and 2. The compounds were evaluated for their antibacterial and antifungal activities against various types of microbial strains, and they showed comparable activity with that of standard drugs. It conclude that, the synthesized compounds are as inhibitor and destructor of microbes, it have strong agent to kill all the microbes and inhibit the growth of microbes.

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